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Hiemenz, P.C.,  
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Sol, in physical chemistry, a colloid (aggregate of very fine particles dispersed in a continuous medium) in which the particles are

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solid and the dispersion medium is fluid. If the dispersion medium is water, the colloid may be called a hydrosol; and if air, an aerosol. Lyophobic (Greek: "liquid-hating") sols are characterized by particles that are not strongly attracted to molecules of ...

**Sol | colloid |  
Britannica**

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Paul C. Hiemenz ,

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Timothy P. Lodge

Written by well-established professors in the field, Polymer Chemistry, Second Edition provides a well-rounded and articulate examination of polymer properties at the molecular level.

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A colloid is a type of homogeneous mixture in which the dispersed

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particles do not settle out. The insoluble particles in the mixture are microscopic, with particle sizes between 1 and 1000 nanometers. The mixture may be termed a colloid or a colloidal suspension. The phrase "colloidal solution" is incorrect.

## **Colloid - Chemistry Glossary Definition - ThoughtCo**

Colloids . Particles

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intermediate in size between those found in solutions and suspensions can be mixed in such a way that they remain evenly distributed without settling out. These particles range in size from  $10^{-8}$  to  $10^{-6}$  m in size and are termed colloidal particles or colloids. The mixture they form is called a colloidal dispersion.

# Where To Download Colloid Chemistry **Solutions, Suspensions, Colloids, and Dispersions**

In chemistry, a colloid is a phase separated mixture in which one substance of microscopically dispersed insoluble or soluble particles is suspended throughout another substance.

Sometimes the dispersed substance alone is called the colloid; the term

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colloidal suspension  
refers unambiguously  
to the overall mixture  
(although a narrower  
sense of the word  
suspension is  
distinguished from  
colloids ...

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## **Polymer Chemistry Hiemenz Solution**

Topic hierarchy A colloid is one of the three primary types of mixtures, with the other two being a solution and



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suspension. A colloid is a mixture that has particles ranging between 1 and 1000 nanometers in diameter, yet are still able to remain evenly distributed throughout the solution.

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A group of mixtures  
called colloids (or  
colloidal dispersions)  
exhibit properties  
intermediate between

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those of suspensions and solutions. The particles in a colloid are larger than most simple molecules; however, colloidal particles are small enough that they do not settle out upon standing. Figure 1.

## **11.5 Colloids - Chemistry**

Colloids, in chemistry, are a mixture of two substances, in which one substance is

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divided into minute particles aka colloidal particles (ranging from 1 to 1000 nm in diameter) and dispersed or suspended over another substance. These insoluble particles are inseparable either by filtering or centrifuging.

**Colloids - Definition,  
Types,  
Classification,  
Application ...**

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20th century witnessed  
various key  
developments in  
physics and chemistry,  
a number of which



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bore directly on colloids. These included advances in the knowledge of the electronic structure of atoms, in the concepts of molecular size and shape, and in insights into the nature of solutions. Moreover, efficient methods for studying the size and configuration of colloidal ...

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